

^{18}F Production

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Outline

Cyclotron technique

^{18}F targetry : Fluoride and Fluorine production



Main PET Isotopes

^{11}C : 20 min, β^+ (970 KeV), $^{14}\text{N}(\text{p},\alpha)^{11}\text{C}$

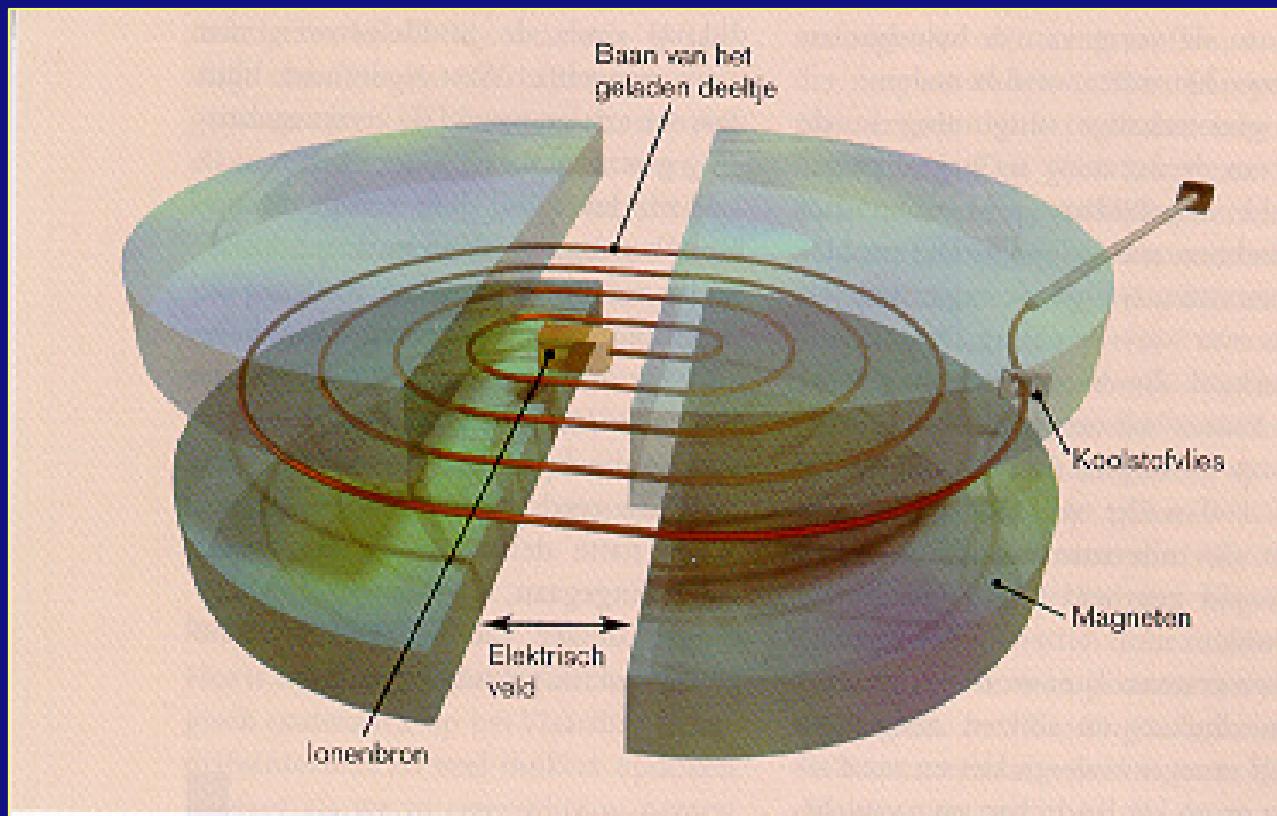
^{13}N : 10 min, β^+ (1190 KeV), $^{16}\text{O}(\text{p},\alpha)^{13}\text{N}$

^{15}O : 2 min, β^+ (1720 KeV), $^{14}\text{N}(\text{d},\text{n})^{15}\text{O}$

^{18}F : 110 min, β^+ (635 KeV), $^{18}\text{O}(\text{p},\text{n})^{18}\text{F}$



Cyclotron production



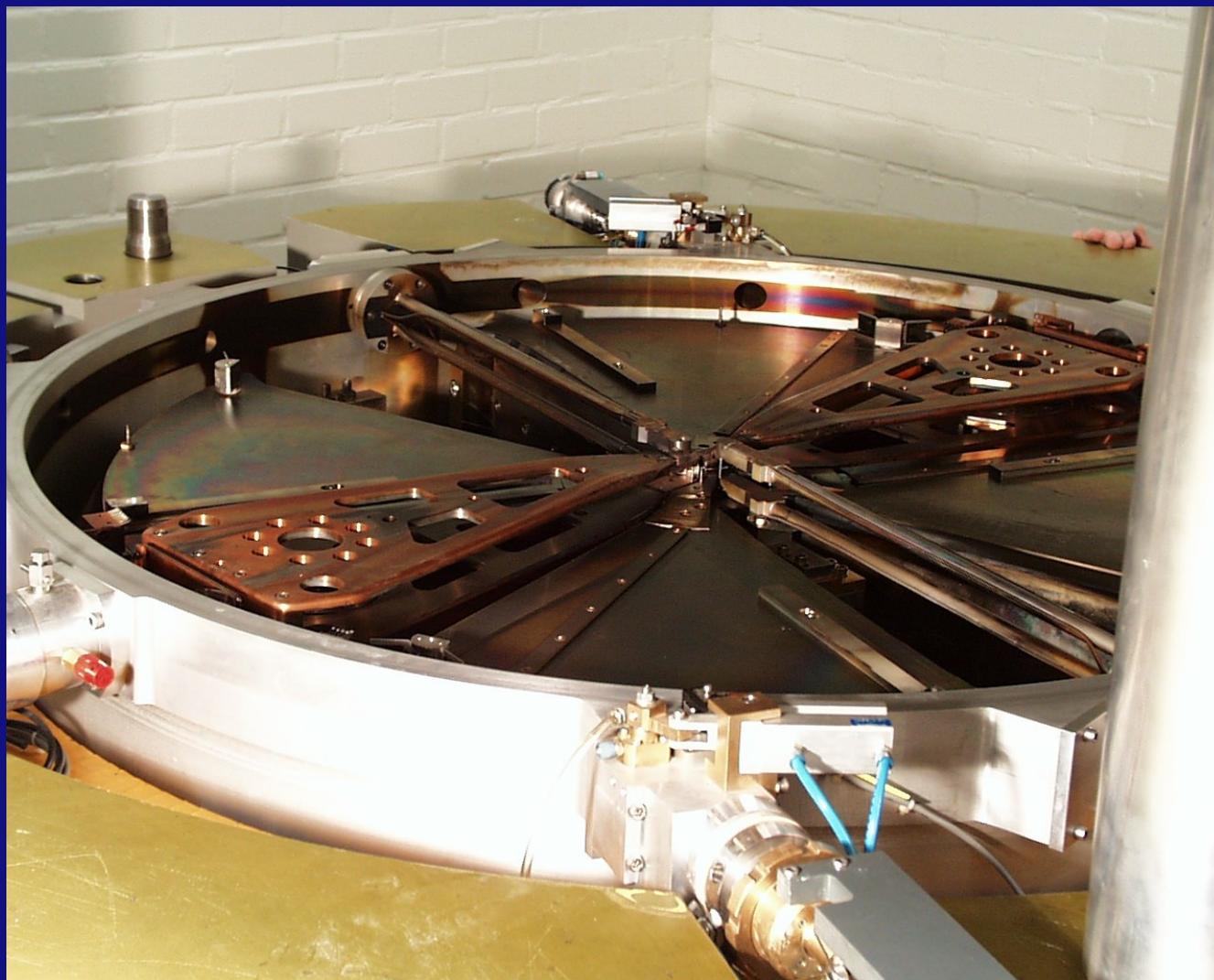
30.000 volt between D's

300 circles, 600 accelerations of 30.000 volt = 18 MeV



Cyclotron





IBA 18/9



^{18}F : 2,3 ml H_2^{18}O in nuobium

^{11}C : 60 ml $\text{N}_2 + 0,5\%$ O_2 in aluminium

^{13}N : 2,3 ml H_2O in nuobium

^{15}O : 60 ml N_2 in aluminium



¹⁸F Fluoride production

Nuclear reaction

$^{18}\text{O}(\text{p},\text{n})^{18}\text{F}$, obtained as Fluoride in water
Proton energy from 10 MeV

Side reaction

$^{16}\text{O}(\text{p},\alpha)^{13}\text{N}$, obtained as nitrates in water

Side products

no nuclear side reactions



^{18}F Fluoride production

Nuclear reaction

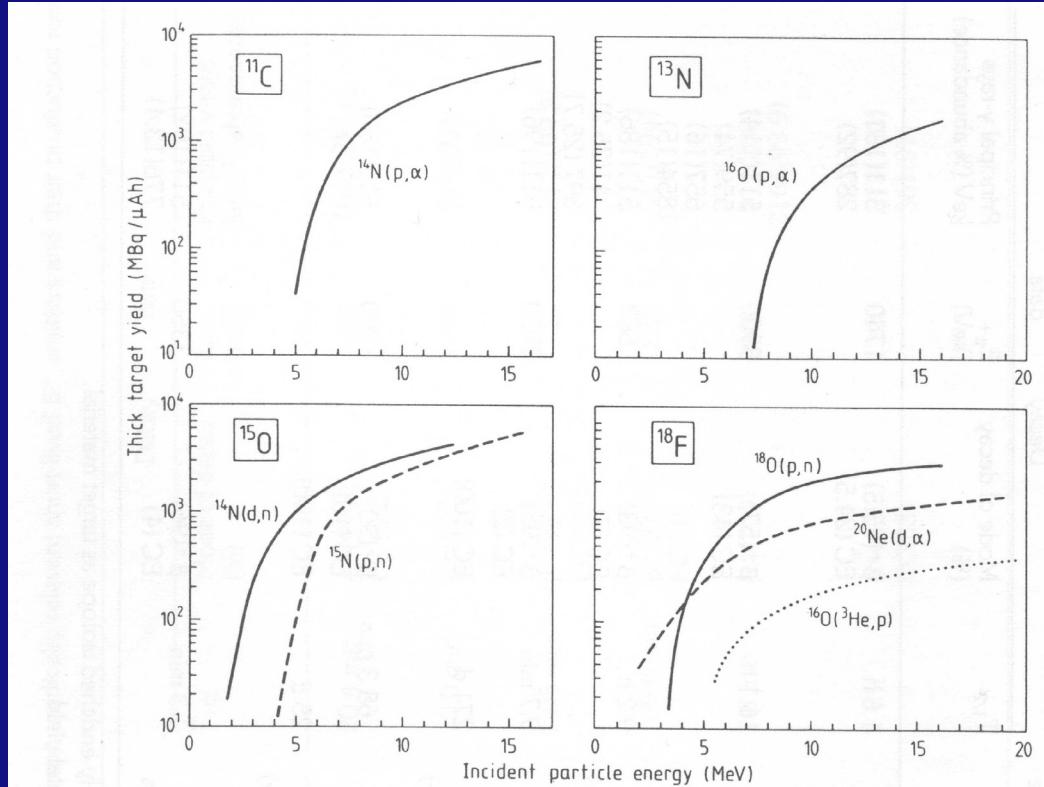
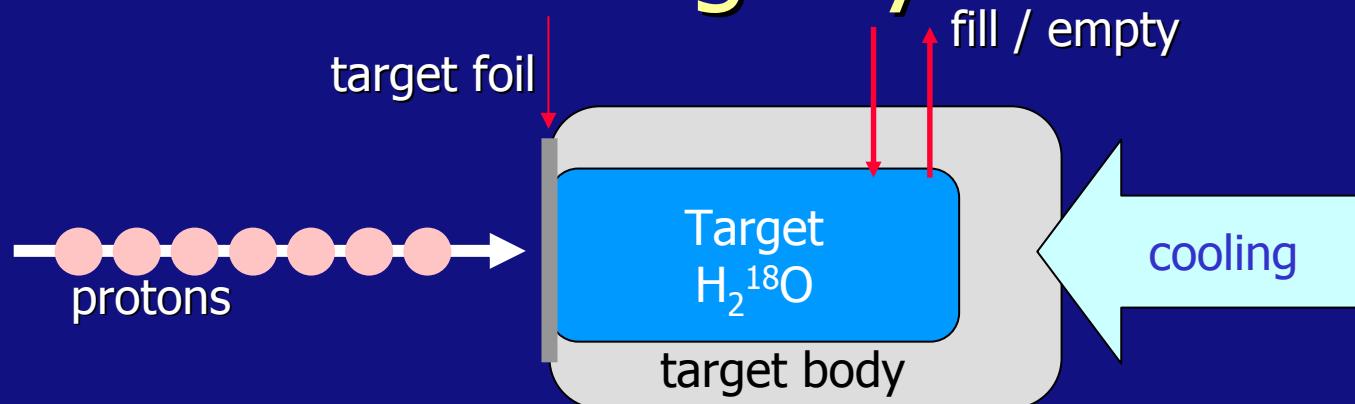


Figure 1. Integrated thick target yields of some commonly used positron-emitters expected from the most common production routes, plotted as a function of incident particle energy. Data were calculated using the measured excitation functions (for reference see text).



^{18}F Fluoride targetry



Important characteristics for target body material:

Heat capacity coefficient

(limiting factor for high beam currents: high pressure in target)

Radionuclidian side products

Chemical purity of $[^{18}\text{F}]\text{Fluoride}$ from target



¹⁸F Fluoride target body material

Silver

Advantage:

- high heat capacity
(max \pm 40 μ A)
- no long lived radio isotopes

Disadvantage:

- Silver particles



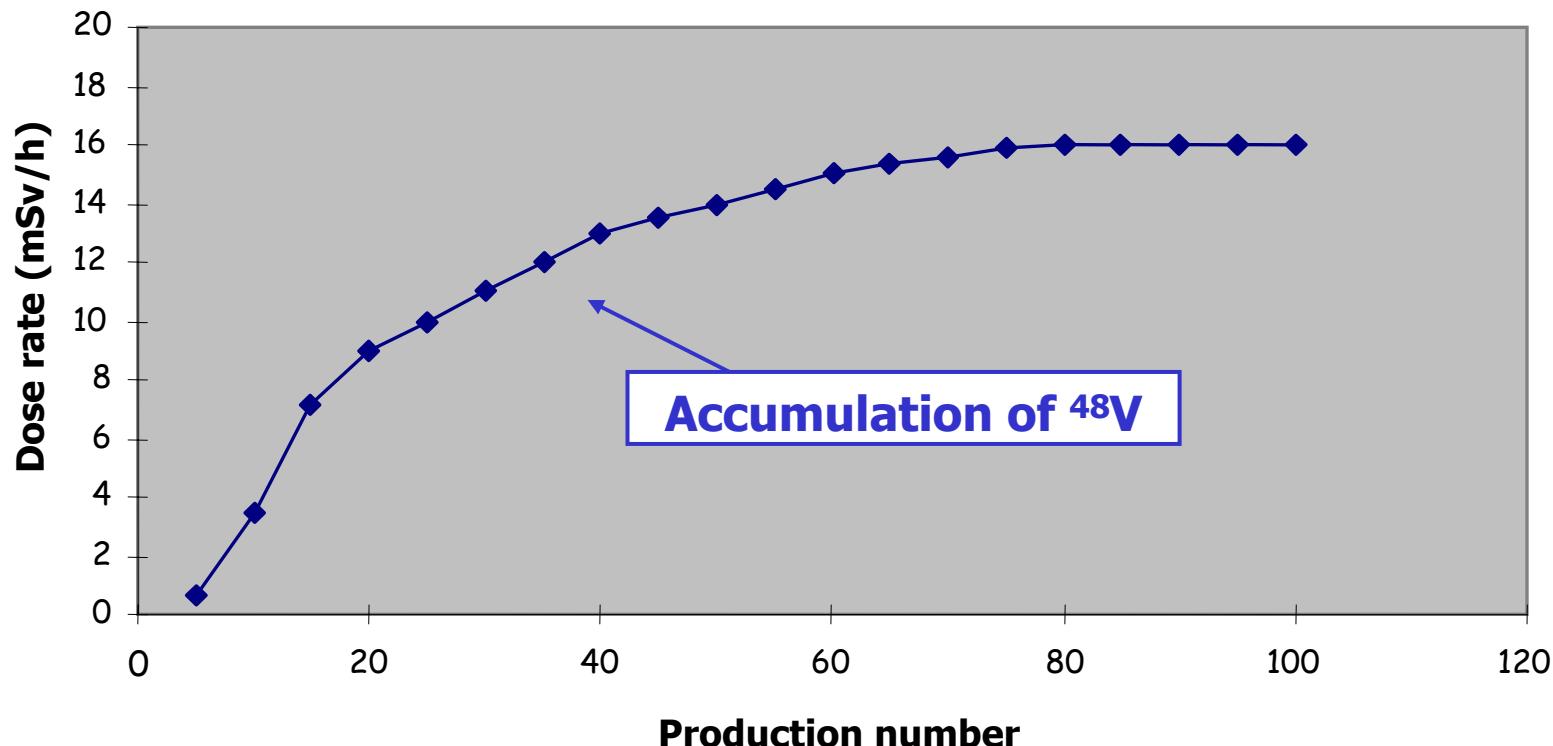
¹⁸F Fluoride target body material

Titanium

- Advantage:
- clean ¹⁸F solution
- Disadvantage:
- low heat capacity
(max \pm 30 μ A)
 - ⁴⁸V as side product



Activation of Titanium



¹⁸F Fluoride target body material

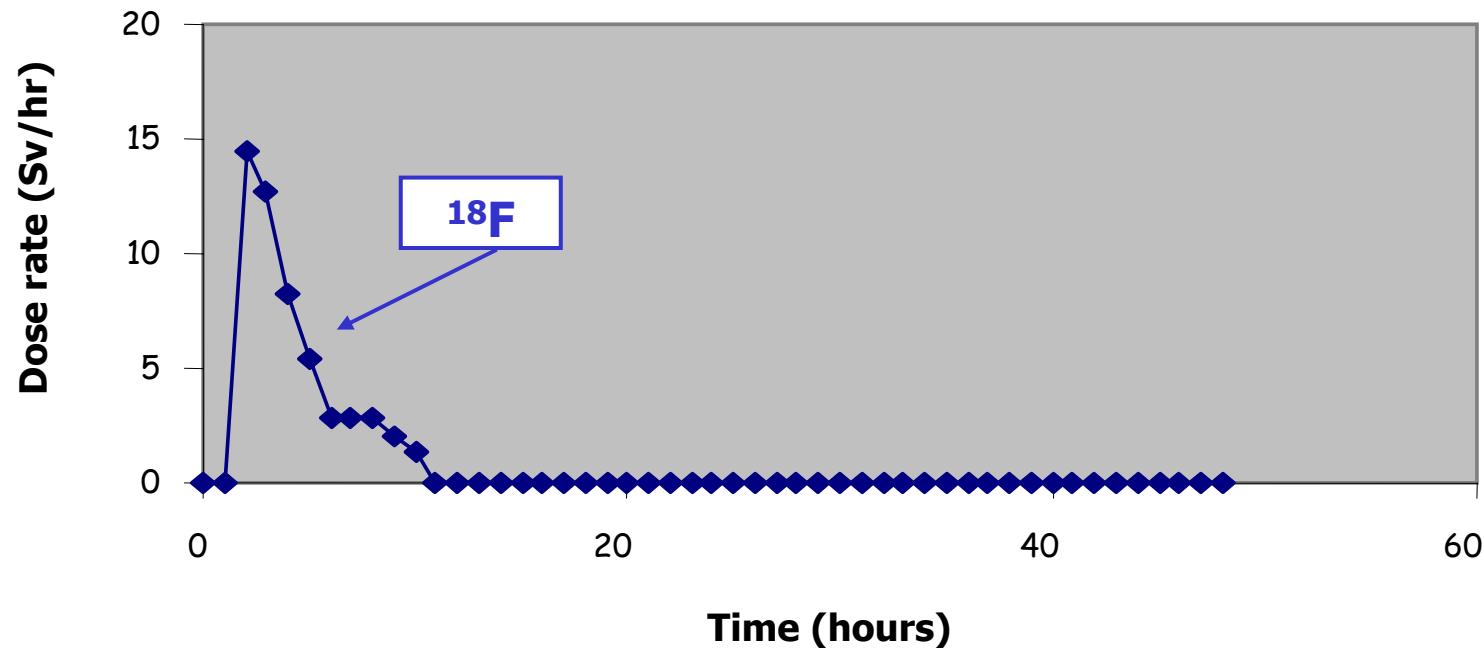
Niobium

- Advantage:
- clean ¹⁸F solution
 - no long lived radio isotopes
 - high heat capacity
(max \pm 40 μ A)

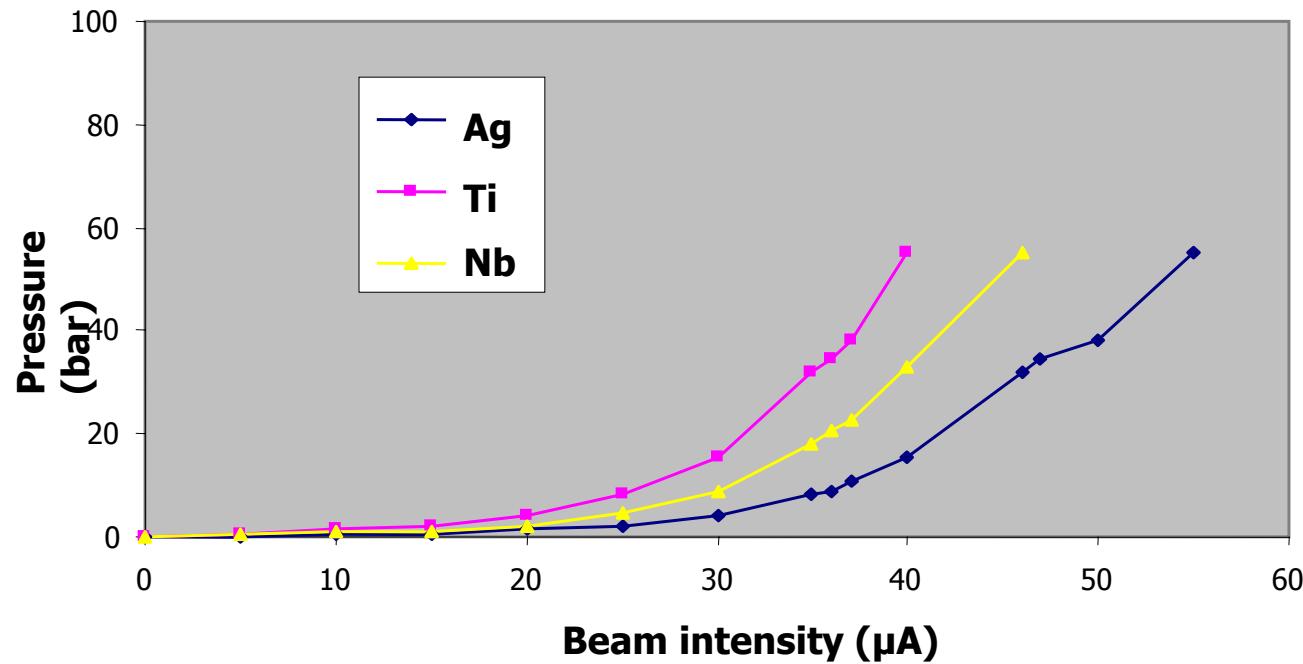
Disadvantage: - ?



Nuobium activation



Pressure in different materials



Results different target materials

Summary of the results after 120 minutes irradiation

Target chamber material	Ag	Ti	Nb
Energy (MeV)	18	18	18
Beam intensity (μ A)	40	30	35
Thick target yield (mCi/ μ A @ sat)	237	245	237
Radioactivity EOB (Ci)	5.1	3.9	4.4



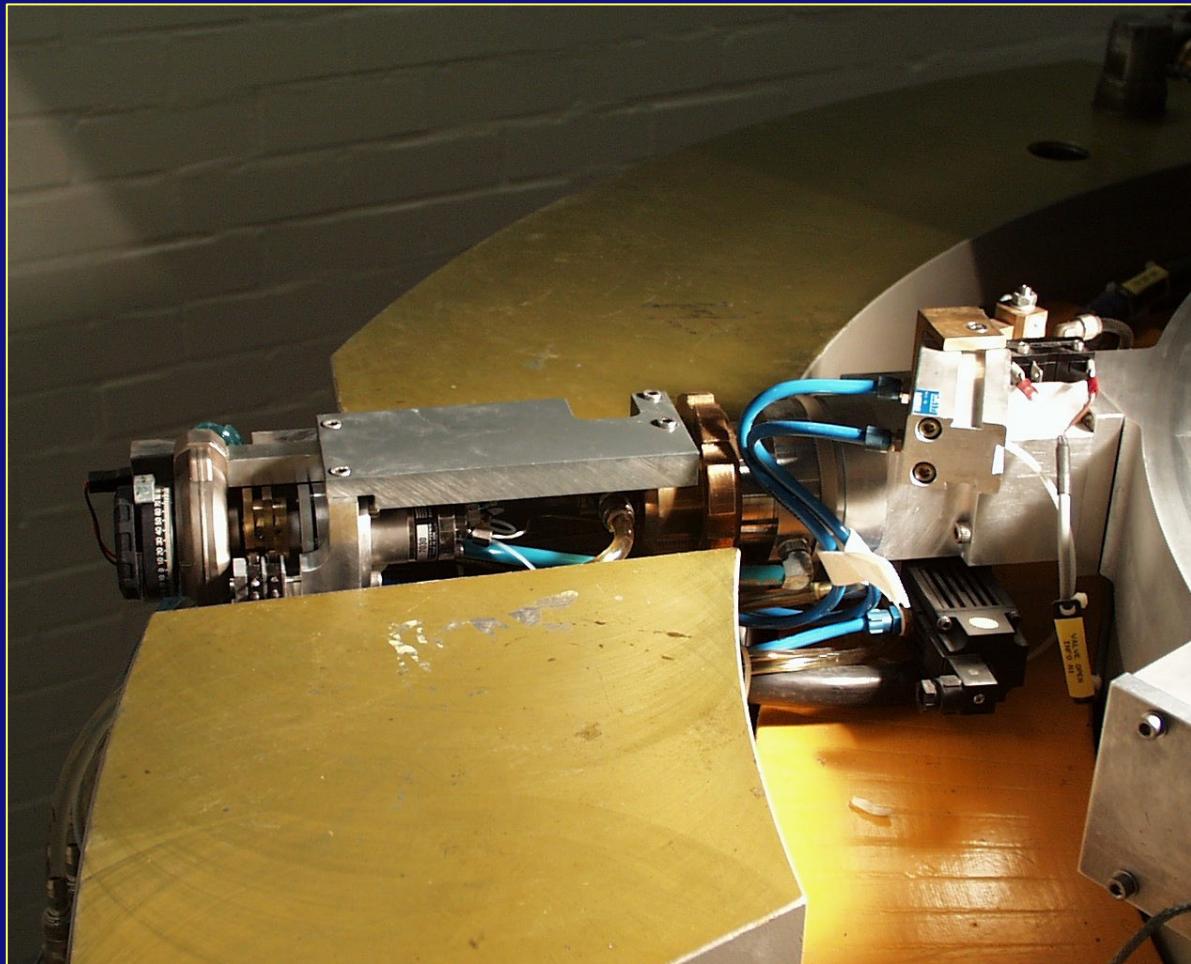
Results at other centers

Table 3. Typical data for [¹⁸F]fluorine produced at different centres from ¹⁸O-enriched water targets.
 (Data reproduced from Guillaume *et al.*, 1991, with permission).

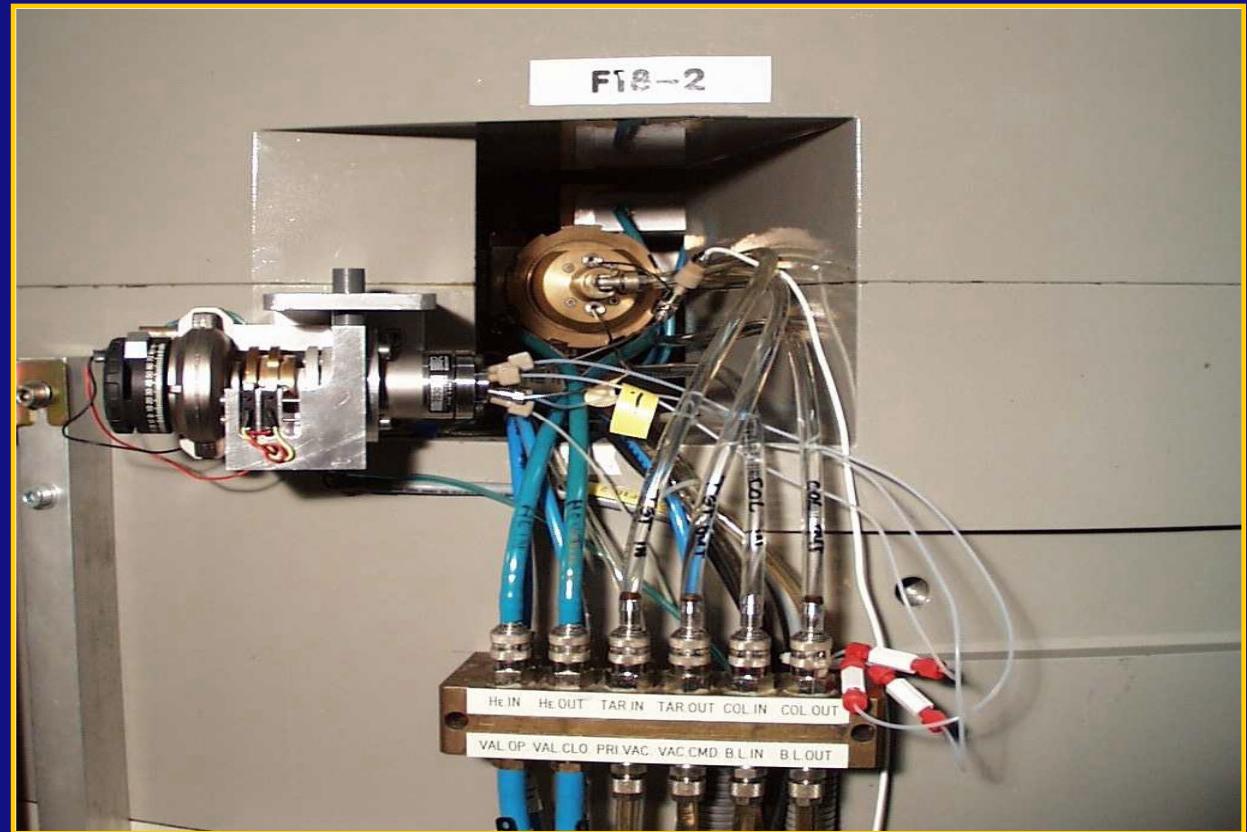
Centre	St. Louis	Sendai	Villigen	Turku	Jülich	Hammersmith	Liège
Material insert	SS & Ti	Ti	Ag	Ag	Ti	SS316 & Ti	Ni
Water width (mm)	3, 5, 7	3, 4, 5	5	1.5	3.5	3	3.4
Foils	Havar or Ti	Al/Ti & Ag	Ag	Ni	Ti	Al/Ti & SS316	Ti
Seal	O-ring	O-ring	Ag	Metal	Weld & Ag	Metal to metal	O-ring
Cooling (°C)	Water (10)	Coolant (0)	Water (30)	Coolant (10)	Water (10)	Water (10)	Water (10)
¹⁸ O-Enrichment of H ₂ O (%)	97	20	98	98	97	20	5
Volume (mL)	1.2, 2.0, 2.8	2.5	4.5	0.195	1.3	2.0	1.8
Pressure (atm)	1	Open or circulated (Pd)	Open	Open	15-25	2	2-3
E (MeV)	15	16	16	12	16.5	16	22.6
Irradiation (μAx min)	15-20x80	20x60	20x60	10x60	20-35x60	20x60	10x60
Yield (GBq/μAh at EOB) (mCi/μAh at EOB)	2.22 <i>ca</i> 60	0.33 9	2.07 56	1.11 30	2.41 70	0.407 11	0.11-0.12 3-3.2
Sp. Act. (TBq/μmol EOB) (Ci/μmol at EOB)	1.85 <i>ca</i> 50	0.148 4	nca	5.18 140	7.4 200	11.1x10 ⁻³ 0.3	0.37 10
Reference	Kilbourn <i>et al.</i> , 1985	Iwata <i>et al.</i> , 1987	Huszár & Weinreich, 1985; Vogt <i>et al.</i> , 1986	Solin <i>et al.</i> , 1988	Qaim <i>et al.</i> , 1987; Nebeling <i>et al.</i> , 1990	Clark <i>et al.</i> , 1990	Guillaume <i>et al.</i> , 1990



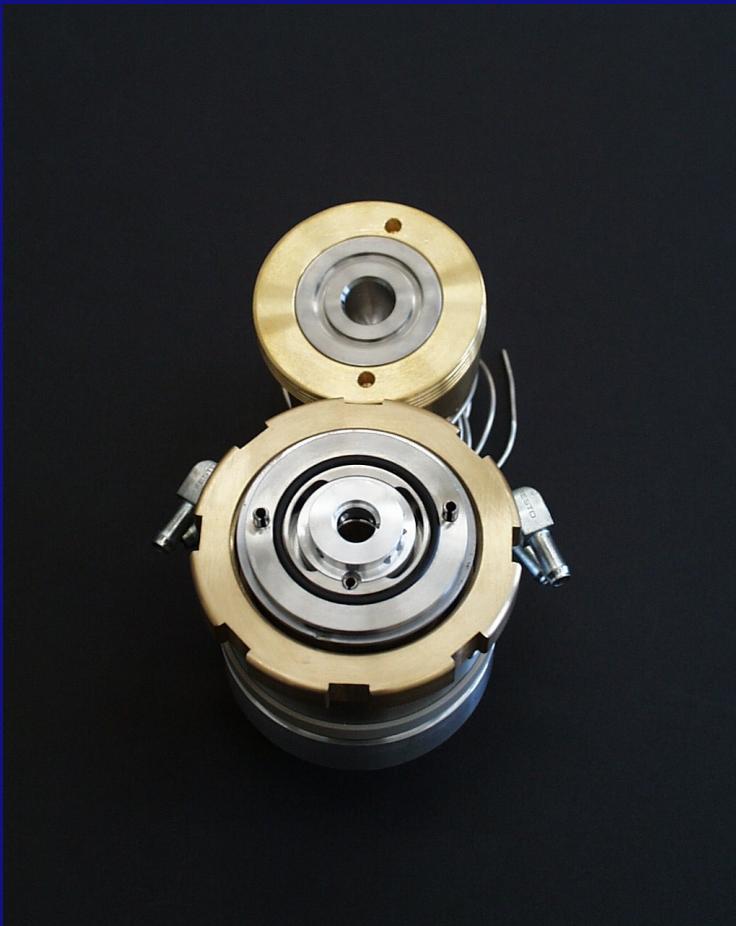
Fluoride target



Fluoride target



Fluoride target



Different materials



^{18}F Fluoride targetry

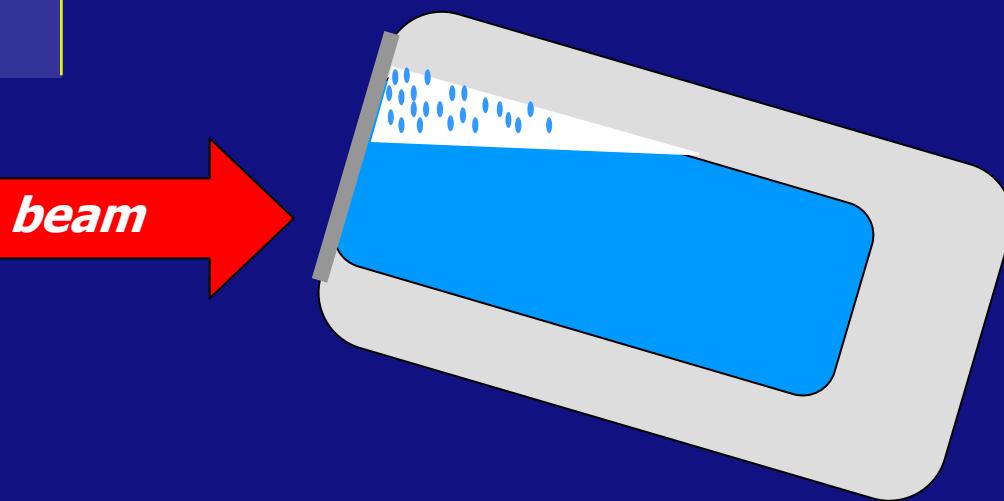
beam →



Target volume $\pm 2.4 \text{ ml}$
Target filling $\pm 2.4 \text{ ml}$
max $40 \mu\text{A}$



^{18}F Fluoride targetry



Target volume $\pm 4.5 \text{ ml}$
Target filling $\pm 3.3 \text{ ml}$
 $70 - 80 \mu\text{A}$

XXXL : $150 \mu\text{A}$?



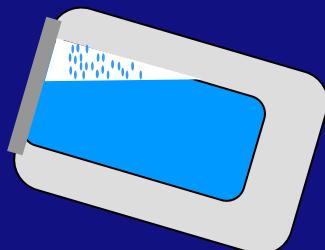
^{18}F and $[^{18}\text{F}]$ FDG yields 2004



3 uur 40 μA

Start synthesis
(End of synthesis)

166 GBq $[^{18}\text{F}]$ Fluoride
130 GBq $[^{18}\text{F}]$ FDG)



3 uur 70 μA

Start synthesis
(End of synthesis)

315 GBq $[^{18}\text{F}]$ Fluoride
240 GBq $[^{18}\text{F}]$ FDG)



¹⁸F Fluorine production

Nuclear reaction

$^{20}\text{Ne}(\text{d},\alpha)^{18}\text{F}$ or $^{18}\text{O}(\text{p},\text{n})^{18}\text{F}$
deuterons from 7 MeV
protons from 10 MeV

Side reaction

-

Side products

-



^{18}F Fluorine targets

Target material : Nickel or Monel

1st irradiation : ^{18}F fluorine deposit on target wall

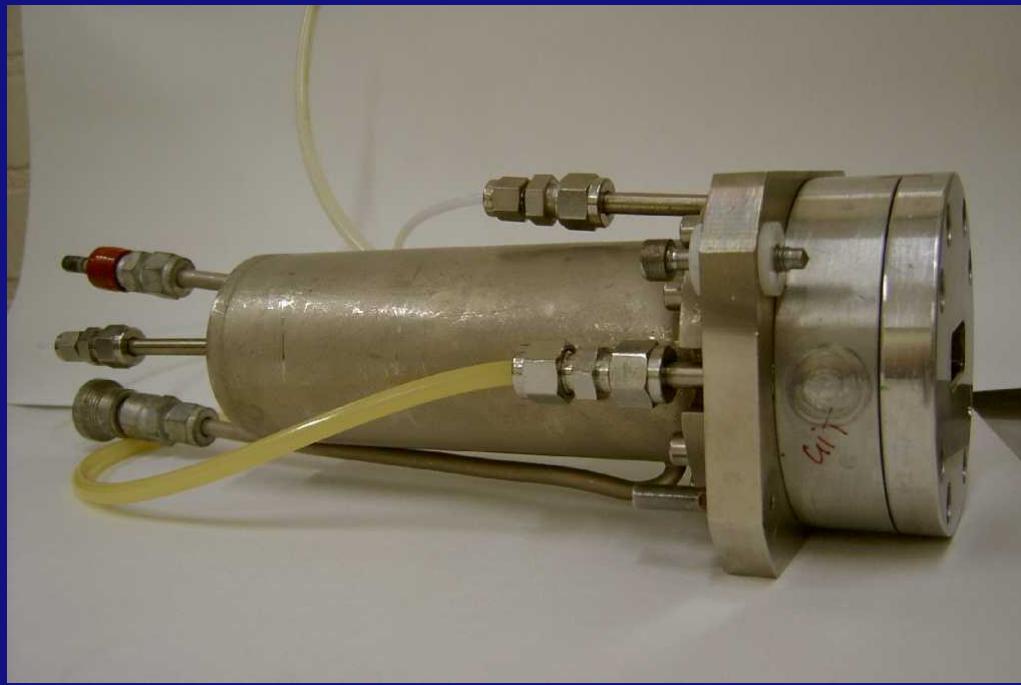
Empty target, fill with Neon + 0.1-0.5% F_2

Irradiate

Electrophilic fluorination, or via intermediate



^{18}F Fluorine target Groningen



Installed in 1992

Design: Robert Dahl

Nickel / 153 ml

30 psi 1% F_2 / 100 psi Neon

Yield in 1992:

5 GBq / 50 μA / 1 hour



^{18}F Fluorine specific radioactivity

Addition of F_2 is essential,
but decreases specific radioactivity

Typically : less than 1 GBq/ μmol

Gives problems for receptor studies,
however for F-DOPA no problem



Production results

Table 2. Typical conditions and production parameters for the cyclotron production of [¹⁸F]fluorine by the ²⁰Ne(d, α) reaction at different centres.
(Data from Guillaume *et al.*, 1991, with permission).

Centre		BNL	MRCCU Hammersmith	Jülich	Liège
Target body	(type; mL) (internal size, cm)	Honed Ni; 50 2.5 i.d. x 10	Polished Ni-201; 100 2.5 i.d. x 20	Ni; 38 2.2 i.d. x 10	Ni; 206
Target gas	(% F ₂ in Ne) ^a (μ mol F ₂) ^b	0.1 50-60	0.12-0.15 88 ^c	0.18 60	0.17 220
Pressure	(beam off, atm) (beam on, atm)	25.8 32.5	13.5 23	18 30	12.3 15
Window	(material, μ m) (seals)	Al, 810-Ni, 25 Metal O-ring	Havar, 50-Ni, 25 Lead	Nb, 25-Havar, 50 Metal joints	Al, 80-Ni, 20 Indium O-rings
E (Incident)	(MeV)	14-9.4	13.8	11.25	12
Irradiation	(μ A x h)	15 x 2	15 x 1.67	40 x 1	10 x 1
Recovery of ¹⁸ F at EOB	(GBq) (mCi)	13.6 367	9.25 250	18.5 500	4.4 120
Yield	(% of theoretical) ^d (MBq/ μ Ah at EOB)	55 463	43 343-370	50 444	51 370-555
Specific activity	(mCi/ μ Ah at EOB) (MBq/ μ mol at EOB)	12.5 259-370	9-10 41	12 130	10-15 29.6-44.4
	(mCi/ μ mol at EOB)	7-10	1.1	3.5	0.8-1.2
Reference		Casella <i>et al.</i> , 1980	Clark <i>et al.</i> , 1990	Blessing <i>et al.</i> , 1986	Guillaume <i>et al.</i> , 1990

^a) Nominal value.

^b) Calculated from target volume and nominal filling parameters.

^c) Experimentally determined as 194-242 μ mol recovered, under conditions in which the fill-line contributes fluorine.

^d) Theoretical yields were calculated according to Casella *et al.* (1980).



Production results Groningen

Radionuclide production:	1998 - 2004	2005
Target	SCX nickel plated / 150 ml Havar frontfoil (25 um)	SCX nickel plated / 150 ml Havar frontfoil (25 um)
Targetgas	0,44 % F2 in Neon / 75 psi	0,44 % F2 in Neon / 75 psi
Beamcurrent	15 uA	25 uA
Irradiation time	60-120 minutes	120 minutes
Preparation for the production	1x conditioning target and lines 1x 3 µAh irradiation to waste	1 x conditioning target and lines 1 x 3 µAh irradiation to waste
Yield [¹⁸ F]F ₂	30-50 MBq/µAh at EOB 1.2 – 2.0 GBq at BOS	70 - 80 MBq/µAh at EOB 3.5 – 4.0 GBq at BOS
Specific activity	4.4 ± 1.0 GBq/mmol	10.7 ± 3.0 GBq/mmol



New developments

Optimization of ^{18}O gas targets : yields 40-50 GBq

Turku method : transfer $[^{18}\text{F}]\text{fluoride}$ to $[^{18}\text{F}]\text{F}_2$

Higher yields and thereby increasing specific activity



Further reading

Radiopharmaceuticals for Positron Emission Tomography

Ed G. Stöcklin and W.W. Pike

Kluwer Academic publishers, Dordrecht, 1993, isbn 0-7923-2340-8

Nuclear Science series:

Fluorine-18 Labeling of Radiopharmaceuticals

M.R. Kilbourn

National Academy Press, Washington, 1990

**Handbook of Radiopharmaceuticals :
Radiochemistry and applications**

Ed MJ Welch, CS Redvanly

Wiley, Sussex, 2003, isbn 0-471-49560-3

